|  |  |  |
| --- | --- | --- |
|  | AP Chemistry | Honors Chemistry |
| Monday | In-service Day | In-service Day |
| Tuesday | 1. Quiz Equilibrium II
2. Hand in Chemical Equilibrium Lab (Pasco 12)
3. **Start Notes & discussion Equilibrium III**
	* **Gibbs Free Energy and Equilibrium**
 | Review 11.1-11.2 Quiz ResultsReview/Hand in Double Replacement Lab1. Start Notes – 11.3 Reactions in Aqueous solutions
	1. Use Double replacement examples
 |
| Wednesday | 1. Review Quiz Equilibrium II
2. **Complete Notes & discussion Equilibrium III**
	* **Gibbs Free Energy and Equilibrium**
3. Complete Problems # 1 - #15 on Equilibrium III worksheet.
 | 1. Finish Notes – 11.3 Reactions in Aqueous solutions
2. Handout sheet HW:: 11.3 Complete & Net ionic
 |
| Thursday | 1. Review/Grade Problems # 1 - #15 on Equilibrium III worksheet.
2. Complete Practice Exam Equilibrium Unit.
 | 1. Grade HW 11.3
2. Lab – SSCL – Precipitation Reactions: Formation of a Solid
3. Complete lab problems at home
 |
| Friday | Exam Equilibrium Unit | 1. Grade 11.3 Packet
2. Complete Chapter 11 Review
 |
|  | * What are we learning? Chemical Equilibrium
* Why are we learning this? One of the 5 big ideas in AP Chemistry
* How will we know when we have learned this? By successfully complete HW and exams.
 | * What are we learning? Types of Chemical Reactions
* Why are we learning this? So TLW have a basic understanding of the types of chemical reactions and there relationship to chemistry processes.
* How will we know when we have learned this? By successfully complete HW and exams.
 |